Synthesis of 3-hydroxy-5-alkoxyhomophthalates by domino '2:1-coupling/intramolecular aldol condensation' reactions of 1,3-bis(trimethylsilyloxy)-1,3-butadienes with tetraalkoxymethanes†

Mathias Lubbe^{*a*} and Peter Langer^{**a,b*}

Received 8th September 2009, Accepted 20th November 2009 First published as an Advance Article on the web 18th December 2009 DOI: 10.1039/b918466j

The first domino '2:1 condensation/intramolecular aldol' reactions of 1,3-bis(trimethylsilyloxy)-1,3-butadiene with tetraalkoxymethanes provide a convenient approach to 3-hydroxy-5-alkoxyhomophthalates. These products, which contain one free and one protected hydroxyl group, can be functionalized by palladium(0)-catalyzed cross-coupling reactions.

Introduction

A great variety of pharmacologically important natural products are biosynthetically derived from poly(β -oxo)carboxylic acids (polyketides, **A**).¹ This includes a variety of polyhydroxylated arenes, which are biogenetically formed by intramolecular aldol condensation of polyketides. Naturally occurring 3hydroxyhomophthalic acid derivatives, such as premithramycinon H (**C**)² and sclerin (**D**),³ are biogenetically derived from triketodicarboxylic acids, such as parent hepta-2,4,6-triketo-1,7dicarboxylic acid **B**, which can be regarded as pentaketides (Scheme 1).



Scheme 1 Polyketides (A), hepta-2,4,6-triketo-1,7-dicarboxylic acid (B) and polyketide-derived homophthalates (C, D).

Harris and coworkers reported the biomimetic synthesis of various polyketides based on condensations of 1,3-dicarbonyl dianions or 1,3,5-tricarbonyl trianions with carboxylic acid derivatives and subsequent spontaneous intramolecular aldol reaction.⁴ In this context, Harris and coworkers reported the formation of 3,5-dihydroxyhomophthalic acid in low yield by reaction of the trianion of heptane-2,4,6-trione with carbon dioxide. The reaction proceeds by formation of hepta-2,4,6-triketo-1,7-dicarboxylic acid **B** and subsequent cyclization.⁵

1,3-Bis(silyloxy)-1,3-butadienes can be regarded as equivalents of 1,3-dicarbonyl dianions.⁶ Recently, we reported the synthesis of 3,5-dioxoesters by condensation of 1,3-bis(silyloxy)-1,3butadienes with acid chlorides.7 In addition, 3,5-dioxopimelic acid diesters, stable 1,3,5,7-tetracarbonyl derivatives, were prepared by reaction of 1,3-bis(silyloxy)-1,3-butadienes with methyl malonyl chloride.8 Chan and Brownbridge have reported the synthesis of homophthalates by 2:1 condensation of 1,3-bis(silyloxy)-1,3butadienes with trimethyl orthoformate or 1,1,1-trimethoxyethane and subsequent cyclization by intramolecular aldol reaction.³ We have reported the synthesis of 4-methoxysalicylates by [3 +3] cyclocondensation of 1,3-bis(silyloxy)-1,3-butadienes with 3oxo-orthoesters.9 Homophthalates have also been prepared by [4 + 2] cycloaddition of 1,3-bis(silyloxy)-1,3-butadienes with dimethyl allene-1,3-dicarboxylate.¹⁰ Herein, we report a convenient approach to 3-hydroxy-5-alkoxyhomophthalates by what are, to the best of our knowledge, the first domino '2:1coupling/intramolecular aldol condensation' reactions of 1,3bis(silyloxy)-1,3-butadienes with tetraalkoxymethanes.

Results and discussion

The TiCl₄-mediated reaction of 1-methoxy-1,3-bis(trimethylsilvloxy)-1,3-butadiene (2a) with tetramethoxymethane (1a) afforded the 3-hydroxy-5-methoxyhomophthalate 3a in 67% yield (Scheme 2). The formation of 3a can be explained by TiCl₄mediated attack of 2a to 1a to give intermediate E, attack of a second molecule of 2a onto E to give intermediate F, subsequent cyclization to give intermediate G and subsequent aromatization (before or during the aqueous work-up). The best yields were obtained when 1a, 2a and TiCl₄ were employed in a stoichiometric ratio 1:2:2, when the reaction was carried out in a relatively concentrated solution (5 mL of CH₂Cl₂ per mmol of 1a) and when hydrochloric acid (10%) was used for the aqueous work up. The use of other Lewis acids, such as Me₃SiOTf, resulted in the formation of complex mixtures. The yield dropped (38%) when 1a, 2a and $TiCl_4$ were employed in a stoichiometric ratio 1:2:1. The yield also dropped (22%) when the reaction was carried out in a more

^aInstitut für Chemie, Universität Rostock, Albert-Einstein-Str. 3a, 18059, Rostock, Germany. E-mail: peter.langer@uni-rostock.de; Fax: 00 49 381 49864112; Tel: 00 49 381 4986410

^bLeibniz-Institut für Katalyse e. V. an der Universität Rostock, Albert-Einstein-Str. 29a, 18059, Rostock, Germany

[†] Electronic supplementary information (ESI) available: Copies of NMR spectra. See DOI: 10.1039/b918466j



Scheme 2 Possible mechanism of the formation of homophthalate 3a.

dilute solution (30 mL of CH_2Cl_2 per mmol of **1a**). No product could be isolated when the reaction was carried out at 0 °C instead of -78 °C.

The condensation of 1,3-bis(trimethylsilyloxy)-1,3-butadienes **2a–g** with tetramethoxymethane (**1a**) and tetraethoxymethane (**1b**) afforded the 3-hydroxy-5-alkoxyhomophthalates **3a–m** in 40–67% yield (Scheme 3, Table 1). The reaction of **1a,b** with dienes containing a terminal substituent resulted in the formation of complex mixtures. The employment of 4-alkyl-1,3-bis(silyloxy)-1,3-butadienes resulted in the formation of complex mixtures. The



Scheme 3 Synthesis of **6a,b**. *Conditions: i*, Tf₂O, pyridine, CH₂Cl₂, $-78 \rightarrow 20$ °C, 4 h; *ii*, **5a,b**, K₃PO₄, Pd(PPh₃)₄, 1,4-dioxane, reflux, 4 h.



^{*a*} Conditions: *i*, (1) **1** (1.0 equiv.), **2a** (2.0 equiv.), TiCl₄ (2.0 equiv.), CH₂Cl₂, $-78 \rightarrow 20$ °C, 14 h; (2) HCl (10%). ^{*b*} Yields of isolated products.

employment of two different dienes in one reaction again resulted in the formation of complex mixtures.

Homophthalates **3** have some synthetic utility because they contain one free and one protected hydroxyl group. For example, derivative **3a** was transformed into its triflate **4**. The Suzuki reaction of **4** with boronic acids **5a,b** afforded the biaryls **6a,b** in high yields (Scheme 3).

Conclusions

The reaction of 1,3-bis(trimethylsilyloxy)-1,3-butadiene with tetraalkoxymethanes afforded 3-hydroxy-5-alkoxyhomophthalates by the first domino '2:1 condensation/intramolecular aldol' reactions. The products, which contain one free and one protected hydroxyl group, could be functionalized by palladium(0)-catalyzed cross-coupling reactions.

Experimental section

General comments

All solvents were dried by standard methods and all reactions were carried out under an inert atmosphere. For ¹H and ¹³C NMR spectra, the deuterated solvents indicated were used. Mass spectrometric data (MS) were obtained by electron ionization (EI, 70 eV), chemical ionization (CI, isobutane) or electrospray ionization (ESI). For preparative scale chromatography, silica gel 60 (0.063–0.200 mm, 70–230 mesh) was used. 1,3-Bis(silyloxy)-1,3-butadienes **2a–g** were prepared according to the literature from the corresponding β -ketoesters in two steps.^{11–13}

General procedure for the synthesis of 3a-m

To a CH₂Cl₂ solution (2 mL per mmol of **1a,b**) of **1a,b** (1.0 mmol) was added **2a–g** (2.0 mmol) and, subsequently, TiCl₄ (0.23 mL, 2.0 mmol) at -78 °C. The temperature of the solution was allowed

to warm to 20 °C over 12–14 h with stirring. To the solution was added HCl (10%, 10 mL), and the organic and aqueous layer were separated. The latter was extracted with CH_2Cl_2 (2 × 10 mL). The combined organic layers were dried (Na₂SO₄) and filtered, and the filtrate was concentrated *in vacuo*. The residue was purified by column chromatography (heptane–EtOAc = 20 : 1).

Dimethyl 3-hydroxy-5-methoxy-homophthalate (3a). Starting with **1a** (0.137 g, 1.0 mmol), **2a** (0.521 g, 2.0 mmol) and TiCl₄ (0.23 mL, 2.0 mmol) in CH₂Cl₂ (5 mL), product **3a** was isolated as a colourless solid (0.170 g, 67%); mp = 77–78 °C; $R_{\rm f}$ 0.70 (heptane–EtOAc = 1 : 1). ¹H NMR (CDCl₃, 250 MHz): δ = 11.64 (s, 1H, OH), 6.46 (d, ⁴J = 2.7 Hz, 1H, CH), 6.29 (d, ⁴J = 2.7 Hz, 1H, CH), 3.85–3.81 (m, 8H, 2 × OCH₃, ArCH₂), 3.68 (s, 3H, OCH₃). ¹³C NMR (75 MHz, CDCl₃): δ = 171.8, 171.0 (C=O), 165.7, 164.1 (C–O), 138.0 (C), 112.8 (CH), 104.9 (C), 100.1 (CH), 55.4, 51.9, 51.7 (OCH₃), 42.7 (ArCH₂). IR (ATR, cm⁻¹): \tilde{v} = 2952 (m), 2845 (m), 1726 (s), 1650 (s), 1621 (s), 1585 (s), 1433 (s), 1386 (m), 1333 (s). MS (EI, 70 eV): *m*/*z* (%) = 254 (M⁺, 54), 223 (23), 222 (57), 195 (42), 194 (88), 190 (25), 180 (12), 179 (100), 165 (10). Anal. Calcd for C₁₂H₁₄O₆ (254.24): C, 56.69; H, 5.55. Found: C, 56.45; H, 5.62.

Diethyl 3-hydroxy-5-methoxy-homophthalate (3b). Starting with 1a (0.137 g, 1.0 mmol), 2b (0.549 g, 2.0 mmol) and TiCl₄ (0.23 mL, 2.0 mmol) in CH₂Cl₂ (5 mL), product **3b** was isolated as a colourless solid (0.119 g, 43%); mp = 64–65 °C; $R_{\rm f}$ 0.70 (heptane–EtOAc = 1 : 1). ¹H NMR (CDCl₃, 250 MHz): δ = 11.78 (s, 1H, OH), 6.42 (d, ${}^{4}J = 2.6$ Hz, 1H, CH), 6.28 (d, ${}^{4}J = 2.6$ Hz, 1H, CH), 4.35 (q, ${}^{3}J = 7.2$ Hz, 2H, OCH₂), 4.14 (q, ${}^{3}J = 7.2$ Hz, 2H, OCH₂), 3.86 (s, 2H, ArCH₂), 3.81 (s, 3H, OCH₃), 1.36 (t, ${}^{3}J =$ 7.2 Hz, 3H, CH_2CH_3), 1.24 (t, ${}^{3}J = 7.2$ Hz, 3H, CH_2CH_3). ${}^{13}C$ NMR (75 MHz, CDCl₃): $\delta = 171.3$, 170.8 (C=O), 165.7, 163.9 (C-O), 138.0 (C), 112.7 (CH), 105.3 (C), 100.1 (CH), 61.4, 60.7 (OCH₂), 55.3 (OCH₃), 42.9 (ArCH₂), 14.2, 14.0 (CH₂CH₃). IR (ATR, cm⁻¹): $\tilde{v} = 2985$ (m), 2940 (m), 2908 (w), 1724 (s), 1641 (s), 1613 (s), 1588 (s), 1447 (m), 1433 (m), 1366 (s). MS (EI, 70 eV): m/z (%) = 282 (M⁺, 36), 237 (22), 236 (41), 208 (29), 190 (21), 181 (41), 180 (100), 179 (21). Anal. Calcd for C₁₄H₁₈O₆ (282.29): C, 59.57; H, 6.43. Found: C, 59.63; H, 6.47.

Di(n-propyl) 3-hydroxy-5-methoxy-homophthalate (3c). Starting with 1a (0.137 g, 1.0 mmol), 2c (0.578 g, 2.0 mmol) and TiCl₄ (0.23 mL, 2.0 mmol) in CH₂Cl₂ (5 mL), product 3c was isolated as a colourless oil (0.143 g, 47%); R_f 0.74 (heptane–EtOAc = 1:1). ¹H NMR (CDCl₃, 300 MHz): $\delta = 11.79$ (s, 1H, OH), 6.42 (d, ⁴J = 2.6 Hz, 1H, CH), 6.28 (d, ${}^{4}J = 2.6$ Hz, 1H, CH), 4.25 (t, ${}^{3}J =$ 6.8 Hz, 2H, OCH₂), 4.03 (t, ${}^{3}J = 6.8$ Hz, 2H, OCH₂), 3.88 (s, 2H, ArCH₂), 3.81 (s, 3H, OCH₃), 1.82–1.57 (m, 4H, 2×CH₂CH₃), 0.99 $(t, {}^{3}J = 7.5 \text{ Hz}, 3\text{H}, \text{CH}_{2}\text{CH}_{3}), 0.90 (t, {}^{3}J = 7.5 \text{ Hz}, 3\text{H}, \text{CH}_{2}\text{CH}_{3}).$ ¹³C NMR (75 MHz, CDCl₃): δ = 171.3, 171.0 (C=O), 165.7, 163.9 (C-O), 138.1 (C), 112.6 (CH), 105.3 (C), 100.1 (CH), 67.2, 66.4 (OCH₂), 55.4 (OCH₃), 42.7 (ArCH₂), 21.9, 21.8 (CH₂CH₃), 10.5, 10.3 (CH₂CH₃). IR (ATR, cm⁻¹): $\tilde{v} = 2967$ (m), 2839 (m), 2880 (w), 1734 (s), 1649 (s), 1615 (s), 1576 (s), 1462 (m), 1430 (m), 1397 (m). MS (EI, 70 eV): m/z (%) = 310 (M⁺, 27), 251 (9), 252 (19), 209 (19), 208 (47), 190 (16), 181 (38), 180 (100). Anal. Calcd for C₁₆H₂₂O₆ (310.14): C, 61.92; H, 7.15. Found: C, 61.84; H, 7.23.

Diisopropyl 3-hydroxy-5-methoxy-homophthalate (3d). Starting with 1a (0.137 g, 1.0 mmol), 2d (0.578 g, 2.0 mmol) and TiCl₄

(0.23 mL, 2.0 mmol) in CH₂Cl₂ (5 mL), product **3d** was isolated as a colourless oil (0.146 g, 47%); $R_{\rm f}$ 0.73 (heptane–EtOAc = 1:1). ¹H NMR (CDCl₃, 250 MHz): δ = 11.90 (s, 1H, OH), 6.41 (d, ⁴J = 2.7 Hz, 1H, CH), 6.26 (d, ⁴J = 2.7 Hz, 1H, CH), 5.27 (septet, ³J = 6.3 Hz, 1H, OCH), 4.99 (septet, ³J = 6.3 Hz, 1H, OCH), 3.86 (s, 2H, ArCH₂), 3.80 (s, 3H, OCH₃), 1.36 (d, ³J = 6.3 Hz, 6H, CH(CH₃)₂), 1.23 (d, ³J = 6.3 Hz, 6H, CH(CH₃)₂). ¹³C NMR (63 MHz, CDCl₃): δ = 170.8, 170.5 (C=O), 165.6, 163.7 (C–O), 138.1 (C), 112.4 (CH), 105.7 (C), 100.1 (CH), 69.6, 68.2 (OCH), 55.3 (OCH₃), 42.9 (ArCH₂), 21.8, 21.7 (CH(CH₃)₂). IR (ATR, cm⁻¹): $\tilde{\nu}$ = 2980 (m), 2838 (m), 2878 (w), 1730 (s), 1647 (s), 1614 (s), 1577 (s), 1466 (m), 1429 (m), 1374 (s). MS (EI, 70 eV): *m*/*z* (%) = 310 (M⁺, 37), 209 (42), 208 (90), 190 (14), 181 (45), 180 (100), 164 (35), 135 (15). Anal. Calcd for C₁₆H₂₂O₆ (310.34): C, 61.92; H, 7.15. Found: C, 61.95; H, 7.23.

Diisobutyl 3-hydroxy-5-methoxy-homophthalate (3e). Starting with 1a (0.137 g, 1.0 mmol), 2e (0.606 g, 2.0 mmol) and TiCl₄ (0.23 mL, 2.0 mmol) in CH₂Cl₂ (5 mL), product 3e was isolated as a colourless oil (0.205 g, 61%); $R_{\rm f}$ 0.72 (heptane–EtOAc = 1:1). ¹H NMR (CDCl₃, 300 MHz): $\delta = 11.78$ (s, 1H, OH), 6.42 (d, ⁴J = 2.6 Hz, 1H, CH), 6.29 (d, ${}^{4}J = 2.6$ Hz, 1H, CH), 4.08 (d, ${}^{3}J =$ 7.0 Hz, 2H, OCH₂), 3.91 (s, 2H, ArCH₂), 3.84 (d, ${}^{3}J = 6.6$ Hz, 2H, OCH₂), 3.81 (s, 3H, OCH₃), 2.06 (septet, ${}^{3}J = 6.8$ Hz, 1H, $CH(CH_{3})_{2}$), 1.90 (septet, ${}^{3}J = 6.8$ Hz, 1H, $CH(CH_{3})_{2}$), 0.98 (d, ${}^{3}J = 6.3$ Hz, 6H, CH(CH₃)₂), 0.88 (d, ${}^{3}J = 6.3$ Hz, 6H, CH(CH₃)₂). ¹³C NMR (75 MHz, CDCl₃): $\delta = 171.2, 171.0$ (C=O), 165.7, 163.9 (C-O), 138.0 (C), 112.5 (CH), 105.4 (C), 100.1 (CH), 71.8, 70.9 (OCH₂), 55.4 (OCH₃), 42.6 (ArCH₂), 27.6, 27.5 (CH(CH₃)₂), 19.3, 19.0 (CH(*C*H₃)₂). IR (ATR, cm⁻¹): $\tilde{v} = 2961$ (s), 2875 (m), 1735 (s), 1649 (s), 1616 (s), 1577 (s), 1467 (m), 1428 (m), 1371 (s). MS (EI, 70 eV): m/z (%) = 338 (M⁺, 32), 209 (38), 208 (95), 190 (12), 182 (11), 181 (45), 180 (100), 164 (20), 135 (12). HRMS (EI, 70 eV): calcd for C₁₈H₂₆O₆ (M⁺) 338.17239, found 338.172288.

Di(2-methoxyethyl) 3-hydroxy-5-methoxy-homophthalate (3f). Starting with 1a (0.137 g, 1.0 mmol), 2f (0.610 g, 2.0 mmol) and TiCl₄ (0.23 mL, 2.0 mmol) in CH₂Cl₂ (5 mL), product 3f was isolated as a colourless oil (0.214 g, 63%); $R_{\rm f}$ 0.23 (heptane-EtOAc = 1:1). ¹H NMR (CDCl₃, 300 MHz): δ = 11.56 (s, 1H, OH), 6.41 (d, ${}^{4}J = 2.6$ Hz, 1H, CH), 6.29 (d, ${}^{4}J = 2.6$ Hz, 1H, CH), 4.42 (t, ${}^{3}J = 4.8$ Hz, 2H, OCH₂), 4.23 (t, ${}^{3}J = 4.6$ Hz, 2H, OCH₂), 3.94 (s, 2H, ArCH₂), 3.80 (s, 3H, OCH₃), 3.67 (t, ${}^{3}J =$ 4.8 Hz, 2H, OCH₂), 3.57 (t, ${}^{3}J$ = 4.6 Hz, 2H, OCH₂), 3.39 (s, 3H, OCH₃), 3.35 (s, 3H, OCH₃). ¹³C NMR (75 MHz, CDCl₃): $\delta =$ 171.4, 170.5 (C=O), 165.5, 164.0 (C-O), 138.1 (C), 112.6 (CH), 105.3 (C), 100.1 (CH), 70.3, 69.8, 64.0, 63.8 (OCH₂), 58.8, 58.7, 55.4 (OCH₃), 42.4 (ArCH₂). IR (ATR, cm⁻¹): $\tilde{v} = 2932$ (m), 2889 (m), 1735 (s), 1650 (s), 1616 (s), 1576 (s), 1434 (s), 1368 (m), 1324 (s). MS (EI, 70 eV): m/z (%) = 342 (M⁺, 22), 267 (6), 266 (20), 208 (17), 180 (24), 164 (12), 135 (10), 59 (100). Anal. Calcd for C₁₆H₂₂O₈ (342.34): C, 56.13; H, 6.48. Found: C, 56.33; H, 6.34.

Dibenzyl 3-hydroxy-5-methoxy-homophthalate (3g). Starting with **1a** (0.137 g, 1.0 mmol), **2g** (0.674 g, 2.0 mmol) and TiCl₄ (0.23 mL, 2.0 mmol) in CH₂Cl₂ (5 mL), product **3g** was isolated as a colourless solid (0.227 g, 56%); mp = 89–90 °C; R_f 0.48 (heptane–EtOAc = 1 : 1). ¹H NMR (CDCl₃, 300 MHz): δ = 11.62 (s, 1H, OH), 7.28–7.15 (m, 10H, 2 × Ph), 6.34 (d, ⁴J = 2.6 Hz, 1H, CH), 6.19 (d, ⁴J = 2.6 Hz, 1H, CH), 5.03 (s, 2H, OCH₂), 4.85 (s, 2H,

OCH₂), 3.78 (s, 2H, ArCH₂CO), 3.71 (s, 3H, OCH₃). ¹³C NMR (63 MHz, CDCl₃): δ = 170.9, 170.4 (C=O), 165.8, 164.1 (C–O), 137.9, 135.8, 135.1 (C), 128.7, 128.6, 128.5, 128.4, 128.2, 112.8 (CH), 105.1 (C), 100.2 (CH), 66.8, 66.3 (OCH₂), 55.4 (OCH₃), 42.8 (ArCH₂CO). IR (ATR, cm⁻¹): $\tilde{\nu}$ = 3008 (m), 2972 (m), 2847 (w), 1729 (s), 1648 (s), 1614 (s), 1575 (s), 1449 (m), 1432 (m), 1380 (s). MS (EI, 70 eV): *m/z* (%) = 406 (M⁺, 16), 209 (10), 181 (7), 180 (100), 92 (6), 91 (100). HRMS (EI, 70 eV): calcd for C₂₄H₂₂O₆ (M⁺) 406.14109, found 406.141134.

Diethyl 5-ethoxy-3-hydroxy-homophthalate (3h). Starting with 1b (0.193 g, 1.0 mmol), 2b (0.549 g, 2.0 mmol) and TiCl₄ (0.23 mL, 2.0 mmol) in CH₂Cl₂ (5 mL), product 3h was isolated as a colourless solid (0.126 g, 47%); mp = 62–63 °C; $R_{\rm f}$ 0.54 (heptane– EtOAc = 1:1). ¹H NMR (CDCl₃, 250 MHz): δ = 11.76 (s, 1H, OH), 6.39 (d, ${}^{4}J = 2.5$ Hz, 1H, CH), 6.27 (d, ${}^{4}J = 2.5$ Hz, 1H, CH), 4.34 (q, ${}^{3}J = 7.1$ Hz, 2H, OCH₂), 4.14 (q, ${}^{3}J = 7.1$ Hz, 2H, OCH_2), 4.03 (q, ${}^{3}J = 7.1$ Hz, 2H, OCH_2), 3.84 (s, 2H, $ArCH_2$), 1.43–1.33 (m, 6H, $2 \times CH_3$), 1.24 (t, ${}^{3}J = 7.1$ Hz, 3H, CH₃). ${}^{13}C$ NMR (63 MHz, CDCl₃): $\delta = 171.4, 170.8$ (C=O), 165.6, 163.3 (C-O), 138.0 (C), 113.0 (CH), 105.1 (C), 100.5 (CH), 63.6, 61.4, 60.6 (OCH₂), 42.8 (ArCH₂), 14.5, 14.1, 14.0 (CH₃). IR (ATR, cm⁻¹): $\tilde{v} = 2983$ (m), 2938 (m), 2896 (w), 1731 (s), 1645 (s), 1614 (s), 1575 (s), 1462 (m), 1445 (m), 1368 (s). MS (EI, 70 eV): m/z (%) = 296 (M⁺, 38), 251 (21), 250 (40), 222 (28), 195 (28), 194 (100), 193 (25), 167 (14). Anal. Calcd for $C_{15}H_{20}O_6$ (296.32): C, 60.80; H, 6.80. Found: C, 60.53; H, 6.77.

Di(n-propyl) 5-ethoxy-3-hydroxy-homophthalate (3i). Starting with **1b** (0.193 g, 1.0 mmol), **2c** (0.578 g, 2.0 mmol) and TiCl₄ (0.23 mL, 2.0 mmol) in CH₂Cl₂ (5 mL), product **3i** was isolated as a colourless solid (0.212 g, 66%); mp = 55–56 °C; $R_{\rm f}$ 0.74 (heptane– EtOAc = 1:1). ¹H NMR (CDCl₃, 300 MHz): δ = 11.78 (s, 1H, OH), 6.40 (d, ${}^{4}J = 2.6$ Hz, 1H, CH), 6.27 (d, ${}^{4}J = 2.6$ Hz, 1H, CH), $4.25 (t, {}^{3}J = 6.8 \text{ Hz}, 2\text{H}, \text{OCH}_{2}), 4.07 - 4.00 (m, 4\text{H}, 2 \times \text{OCH}_{2}),$ 3.87 (s, 2H, ArCH₂), 1.82–1.56 (m, 4H, $2 \times CH_2CH_3$), 1.40 (t, ${}^{3}J =$ 7.0 Hz, CH_2CH_3), 0.99 (t, ${}^{3}J = 7.5$ Hz, 3H, CH_2CH_3), 0.90 (t, ${}^{3}J =$ 7.5 Hz, 3H, CH₂CH₃). ¹³C NMR (75 MHz, CDCl₃): $\delta = 171.3$, 171.0 (C=O), 165.7, 163.3 (C-O), 138.0 (C), 113.0 (CH), 105.1 (C), 100.5 (CH), 67.1, 66.4, 63.7 (OCH₂), 42.7 (ArCH₂), 21.9, 21.8 (CH_2CH_3) , 14.5, 10.5, 10.3 (CH_2CH_3) . IR (ATR, cm^{-1}) : $\tilde{v} = 2967$ (m), 2838 (m), 2876 (w), 1730 (s), 1651 (s), 1615 (s), 1576 (s), 1459 (m), 1433 (m), 1406 (m). MS (EI, 70 eV): m/z (%) = 324 (M⁺, 23), 265 (8), 264 (16), 223 (16), 222 (47), 195 (29), 194 (100), 167 (12). Anal. Calcd for C₁₇H₂₄O₆ (324.37): C, 62.95; H, 7.46. Found: C, 62.70; H, 7.53.

Disopropyl 5-ethoxy-3-hydroxy-homophthalate (3k). Starting with **1b** (0.193 g, 1.0 mmol), **2d** (0.578 g, 2.0 mmol) and TiCl₄ (0.23 mL, 2.0 mmol) in CH₂Cl₂ (5 mL), product **3k** was isolated as a colourless oil (0.151 g, 47%); R_f 0.74 (heptane–EtOAc = 1 : 1). ¹H NMR (CDCl₃, 250 MHz): δ = 11.89 (s, 1H, OH), 6.38 (d, ⁴J = 2.6 Hz, 1H, CH), 6.25 (d, ⁴J = 2.6 Hz, 1H, CH), 5.27 (septet, ³J = 6.3 Hz, 1H, OCH), 4.99 (septet, ³J = 6.3 Hz, 1H, OCH), 4.03 (q, ³J = 7.0 Hz, OCH₂), 3.85 (s, 2H, ArCH₂), 1.42–1.34 (m, ³J = 6.3 Hz, ³J = 7.0 Hz, 9H, CH₂CH₃, CH(CH₃)₂), 1.22 (d, ³J = 6.3 Hz, 6H, CH(CH₃)₂). ¹³C NMR (63 MHz, CDCl₃): δ = 170.8, 170.5 (C=O), 165.6, 163.1 (C–O), 138.1 (C), 112.8 (CH), 105.5 (C), 100.4 (CH), 69.6, 68.2 (OCH), 63.6 (OCH₂), 42.9 (ArCH₂), 21.8, 21.7 (CH(CH₃)₂), 14.5 (CH₂CH₃). IR (ATR, cm⁻¹): $\tilde{\nu}$ = 2980

(m), 2838 (m), 2878 (w), 1730 (s), 1647 (s), 1614 (s), 1577 (s), 1466 (m), 1429 (m), 1374 (s). MS (EI, 70 eV): m/z (%) = 324 (M⁺, 35), 223 (44), 222 (94), 196 (17), 195 (34), 194 (100), 178 (28), 167 (16). Anal. Calcd for $C_{17}H_{24}O_6$ (323.37): C, 62.95; H, 7.46. Found: C, 63.03; H, 7.61.

Diisobutyl 5-ethoxy-3-hydroxy-homophthalate (31). Starting with **1b** (0.193 g, 1.0 mmol), **2e** (0.606 g, 2.0 mmol) and TiCl₄ (0.23 mL, 2.0 mmol) in CH₂Cl₂ (5 mL), product 31 was isolated as a colourless oil (0.139 g, 40%); R_f 0.76 (heptane–EtOAc = 1:1). ¹H NMR (CDCl₃, 300 MHz): $\delta = 11.77$ (s, 1H, OH), 6.40 (d, ${}^{4}J = 2.6$ Hz, 1H, CH), 6.28 (d, ${}^{4}J = 2.6$ Hz, 1H, CH), 4.09–4.00 (m, 4H, $2 \times OCH_2$), 3.91 (s, 2H, ArCH₂), 3.83 (d, ${}^{3}J = 6.8$ Hz, 2H, OCH_2), 2.06 (septet, ${}^{3}J = 6.8$ Hz, 1H, $CH(CH_3)_2$), 1.87 (septet, ${}^{3}J = 6.8$ Hz, 1H, CH(CH₃)₂), 1.40 (t, ${}^{3}J = 7.0$ Hz, 3H, CH₂CH₃) 0.98 (d, ${}^{3}J = 6.8$ Hz, 6H, CH(CH₃)₂), 0.88 (d, ${}^{3}J = 6.8$ Hz, 6H, CH(CH₃)₂). ¹³C NMR (75 MHz, CDCl₃): $\delta = 171.2, 171.0$ (C=O), 165.7, 163.9 (C–O), 138.0 (C), 112.9 (CH), 105.2 (C), 100.5 (CH), 71.8, 71.0, 63.7 (OCH₂), 42.6 (ArCH₂), 27.6, 27.6 (CH(CH₃)₂), 19.3, 19.0, 14.5 (CH₃). IR (ATR, cm⁻¹): $\tilde{v} = 2961$ (s), 2875 (m), 1736 (s), 1650 (s), 1617 (s), 1576 (s), 1469 (m), 1422 (m), 1325 (s). MS (EI, 70 eV): m/z (%) = 352 (M⁺, 29), 223 (34), 222 (90), 195 (35), 194 (100), 178 (16), 167 (13). Anal. Calcd for $C_{19}H_{28}O_6$ (352.42): C, 64.75; H, 8.01. Found: C, 64.40; H, 8.18.

Di(2-methoxyethyl) 5-ethoxy-3-hydroxy-homophthalate (3m). Starting with 1b (0.193 g, 1.0 mmol), 2f (0.610 g, 2.0 mmol) and TiCl₄ (0.23 mL, 2.0 mmol) in CH₂Cl₂ (5 mL), product **3m** was isolated as a colourless oil (0.145 g, 41%); R_f 0.36 (heptane/EtOAc = 1 : 1). ¹H NMR (CDCl₃, 300 MHz): δ = 11.54 (s, 1H, OH), 6.38 (d, ${}^{4}J = 2.6$ Hz, 1H, CH), 6.28 (d, ${}^{4}J = 2.6$ Hz, 1H, CH), 4.42 (t, ${}^{3}J = 4.8$ Hz, 2H, OC H_{2} CH₂), 4.23 (t, ${}^{3}J = 4.7$ Hz, 2H, OC H_2 CH₂), 4.03 (q, ${}^{3}J = 7.1$ Hz, 2H, OC H_2 CH₃), 3.93 (s, 2H, ArCH₂), 3.67 (t, ${}^{3}J = 4.8$ Hz, 2H, OCH₂CH₂), 3.57 (t, ${}^{3}J = 4.7$ Hz, 2H, OCH₂CH₂), 3.39 (s, 3H, OCH₃), 3.35 (s, 3H, OCH₃), 1.39 (t, ${}^{3}J = 7.1$ Hz, 2H, OCH₂CH₃). 13 C NMR (63 MHz, CDCl₃): $\delta =$ 171.4, 170.5 (C=O), 165.5, 163.4 (C-O), 138.0 (C), 113.1 (CH), 105.1 (C), 100.5 (CH), 70.3, 69.8, 63.9, 63.8, 63.7 (OCH₂), 58.8, 58.7 (OCH₃), 42.5 (ArCH₂), 14.5 (CH₂CH₃). IR (ATR, cm⁻¹): $\tilde{v} =$ 2981 (m), 2931 (m), 2886 (m), 2820 (m), 1736 (s), 1650 (s), 1616 (s), 1575 (s), 1449 (s), 1370 (m), 1323 (s). MS (EI, 70 eV): m/z $(\%) = 356 (M^+, 29), 280 (21), 222 (22), 294 (33), 178 (11), 150 (4),$ 121 (8), 59 (100). Anal. Calcd for $C_{17}H_{24}O_8$ (356.37): C, 57.30; H, 6.79. Found: C, 57.24; H, 6.70.

Dimethyl 5-methoxy-3-trifluormethansulfonyloxy-homophtalate (4). To a stirred dichloromethane solution (30 mL) of **3a** (0.763 g, 3.0 mmol) was added pyridine (0.475 g, 6.0 mmol) at -78 °C and after stirring for 10 min was dropwise added trifluoromethanesulfonic acid anhydride (1.016 g, 3.6 mmol) at the same temperature. The temperature of the mixture was allowed to warm to 0 °C and then stirred for additional 4 h. After direct purification by column chromatography (CH₂Cl₂), product **4** was isolated as a colourless solid (1.086 g, 94%); *R*_f 0.44 (CH₂Cl₂). ¹H NMR (CDCl₃, 300 MHz): $\delta = 6.83$ (d, ⁴*J* = 2.4 Hz, 1H, CH), 6.75 (d, ⁴*J* = 2.7 Hz, 1H, CH), 3.89–3.88 (m, 5H, OCH₃, ArCH₂), 3.85 (s, 3H, OCH₃), 3.68 (s, 3H, OCH₃). ¹³C NMR (63 MHz, CDCl₃): $\delta = 170.6$, 164.7 (C=O), 161.7 (C–O), 148.9, 138.0 (C), 118.5 (q, *J*_{CF} = 320 Hz, CF₃), 118.4 (C), 117.1, 107.3 (CH), 55.9, 52.4, 52.1 (OCH₃), 39.8 (ArCH₂). ¹⁹F NMR (CDCl₃, 235 MHz): $\delta = -73.6$ (CF₃).

General procedure for the synthesis of 6a,b

A 1,4-dioxane solution (2.5 mL per mmol of 4) of aryl boronic acid **5a,b** (1.3 mmol), K_3PO_4 (1.6 mmol), $Pd(PPh_3)_4$ (3 mol%) and triflate 4 (1.0 mmol) was stirred at 110 °C for 4 h. After cooling to 20 °C, a saturated aqueous solution of NH₄Cl was added. The organic and aqueous layer were separated and the latter was extracted with diethyl ether. The combined organic layers were dried (Na₂SO₄), filtered and the filtrate concentrated *in vacuo*. The residue was purified by column chromatography (heptane–EtOAc = 10:1).

Dimethyl 5-methoxy-3-phenyl-homophthalate (6a). Starting with **4** (0.387 g, 1.0 mmol), **5a** (0.159 g, 1.3 mmol), K₃PO₄ (0.340 g, 1.6 mmol) and Pd(PPh₃)₄ (0.035 g, 0.03 mmol) in 1,4-dioxane (2.5 mL), product **6a** was isolated as a light-yellow solid (0.308 g, 98%); mp = 71–72 °C; R_f 0.58 (heptane–EtOAc = 1 : 1). ¹H NMR (CDCl₃, 300 MHz): δ = 7.42–7.30 (m, 5H, Ph), 6.83 (br s, 2H, 2×CH), 3.85–3.81 (m, 5H, OCH₃, ArCH₂), 3.69 (s, 3H, OCH₃), 3.47 (s, 3H, OCH₃). ¹³C NMR (75 MHz, CDCl₃): δ = 171.3, 169.6 (C=O), 160.1 (C–O), 143.5, 141.7, 134.7 (C), 128.2, 128.0, 127.4 (CH), 125.2 (C), 115.6, 114.3 (CH), 55.4, 51.9, 51.7 (OCH₃), 42.7 (ArCH₂). IR (ATR, cm⁻¹): $\tilde{\nu}$ = 2947 (m), 2848 (m), 1731 (s), 1709 (s), 1602 (s), 1574 (m), 1433 (s), 1313 (m), 1286 (s). MS (EI, 70 eV): *m/z* (%) = 314 (M⁺, 40), 283 (28), 282 (28), 255 (58), 254 (93), 240 (17), 239 (100). Anal. Calcd for C₁₈H₁₈O₅ (314.33): C, 68.78; H, 5.77. Found: C, 68.82; H, 5.72.

Dimethyl 5-methoxy-3-(4-tolyl)-homophthalate (6b). Starting with **4** (0.271 g, 0.70 mmol), **5b** (0.124 g, 0.91 mmol), K_3PO_4 (0.238 g, 1.6 mmol) and Pd(PPh₃)₄ (0.025 g, 0.02 mmol) in 1,4-dioxane (2.0 mL), product **6b** was isolated as a light-yellow solid (0.210 g, 92%); mp = 69–70 °C; R_f 0.60 (heptane–EtOAc = 1 : 1). ¹H NMR (CDCl₃, 300 MHz): δ = 7.24–7.17 (m, 4H, 4×CH), 6.82–6.80 (m, 2H, 2×CH), 3.84 (s, 3H, OCH₃), 3.79 (s, 2H, ArCH₂), 3.69 (s, 3H, OCH₃), 3.51 (s, 3H, OCH₃), 2.38 (s, 3H, ArCH₃). ¹³C NMR (75 MHz, CDCl₃): δ = 171.3, 169.7 (C=O), 160.1 (C–O), 143.4, 138.3, 137.1, 134.5 (C), 128.9, 127.9 (CH), 125.3 (C), 115.4,

114.3 (CH), 55.4, 52.0, 51.7 (OCH₃), 39.4 (ArCH₂), 21.1 (ArCH₃). IR (ATR, cm⁻¹): $\tilde{\nu} = 2948$ (m), 2847 (m), 1731 (s), 1708 (s), 1599 (s), 1574 (m), 1430 (s), 1314 (m), 1283 (s). MS (EI, 70 eV): *m/z* (%) = 328 (M⁺, 53), 297 (28), 296 (20), 269 (59), 268 (93), 254 (19), 253 (100). Anal. Calcd for C₁₉H₂₀O₅ (328.36): C, 69.50; H, 6.14. Found: C, 69.53; H, 6.21.

Acknowledgements

Financial support by the State of Mecklenburg-Vorpommern is gratefully acknowledged.

Notes and references

- (a) J. Staunton and K. J. Weissman, *Nat. Prod. Rep.*, 2001, **18**, 380;
 (b) A. M. P. Koskinen and K. Karisalmi, *Chem. Soc. Rev.*, 2005, **34**, 677.
- 2 K. Krohn and J. Vitz, Eur. J. Org. Chem., 2004, 209.
- 3 T.-H. Chan and P. Brownbridge, J. Chem. Soc., Chem. Commun., 1981, 20.
- 4 For a review, see: (a) T. M. Harris and C. M. Harris, *Tetrahedron*, 1977, 33, 2159; see also: (b) T. P. Murray and T. M. Harris, J. Am. Chem. Soc., 1972, 94, 8253; (c) C. M. Harris, J. S. Roberson and T. M. Harris, J. Am. Chem. Soc., 1976, 98, 5380; (d) T. M. Harris and J. V. Hay, J. Am. Chem. Soc., 1977, 99, 1631; (e) J. S. Hubbard and T. M. Harris, *Tetrahedron Lett.*, 1978, 19, 4601; (f) R. M. Sandifer, A. K. Bhattacharya and T. M. Harris and T. M. Harris, J. Org. Chem., 1981, 46, 2260; (g) S. G. Gilbreath, C. M. Harris and T. M. Harris, J. Am. Chem. Soc., 1988, 110, 6172.
- 5 T. M. Harris, P. M. Murphy and A. J. Poje, J. Am. Chem. Soc., 1976, 98, 7733.
- 6 For a review of 1,3-bis(silyl enol ethers), see: P. Langer, *Synthesis*, 2002, 441.
- 7 (a) T. Rahn, V. T. H. Nguyen, T. H. T. Dang, Z. Ahmed, M. Lalk, C. Fischer, A. Spannenberg and P. Langer, *J. Org. Chem.*, 2007, **72**, 1957; (b) T. Rahn, T. H. T. Dang, A. Spannenberg, C. Fischer and P. Langer, *Org. Biomol. Chem.*, 2008, **6**, 3366.
- 8 S. Reim, D. Michalik, K. Weisz, Z. Xiao and P. Langer, Org. Biomol. Chem., 2008, 6, 3079.
- 9 M. Lubbe, J.-P. Gütlein, H. Reinke and P. Langer, Synlett, 2008, 2671.
- 10 I. Hussain, M. A. Yawer, B. Appel, M. Sher, A. Mahal, A. Villinger and P. Langer, *Tetrahedron*, 2008, 64, 8003.
- 11 T. H. Chan and P. Brownbridge, J. Am. Chem. Soc., 1980, 102, 3534.
 12 C. A. Malardan and K. O. Chang, J. M. Chem. Soc., 1980, 117
- 12 G. A. Molander and K. O. Cameron, J. Am. Chem. Soc., 1993, 115, 830.
- 13 V. T. H. Nguyen, E. Bellur and B. Appel, Synthesis, 2006, 2865.